

## NC-16112020701010200 Seat No. \_\_\_\_

## Master of Pharmacy Management (Sem. I) (CBCS) Examination

January - 2017

## Pharmaceutical Chemistry - I

(Pharmaceutical Inorganic & Physical Chemistry) (New Syllabus)

Time: 3 Hours] [Total Marks: 80

**Instructions**: (1) Attempt three questions from each section.

- Questions 1 and 5 are compulsory. (2)
- (3) Tie each section separately.
- Figure to the right indicates full marks for the (4) respective question.

## SECTION - I

1 Explain the following terms : (any seven) 14 (1) Pharmacopoeia (2) **Impurity** Radioisotope (3) **(4)** Buffer capacity **(5)** Adsorbent (6) Hematinics Respiratory stimulant (7) (8) **Dentifrices** Order of reaction (9)(10) Astringent 2 Discuss sources of impurities in detail. 13 3 (1) Write a note on physiological acid-base balance. 7 (2) What are gastrointestinal agents? Briefly classify 6 them with suitable examples.

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4	Answer the following:		
	(1)	Write down various clinical applications of radiopharmaceuticals.	7
	(2)	Explain solubility chart as well as storage conditions of official substances.	6
		SECTION - II	
5	Answer the following questions: (any two)		14
	(1)	What is catalysis? Write a note on homogeneous and heterogeneous catalyst. Explain important characteristics of enzyme catalysis.	
	(2)	Define limit test. Explain limit test of chloride in detail.	
	(3)	Classify: Dental products. Discuss Sodium fluoride as dental product.	
6	(1)	Define: Antidotes. Discuss mechanism of action of antidote poisoning. Write a note on cyanide poisoning and its treatment.	7
	(2)	What are antioxidants and preservatives? Write preparation and properties of any one antioxidant.	6
7	(1)	Discuss the physiological role of oxygen and describe its method of preparation, properties, storage conditions and uses.	7
	(2)	Differentiate between:	6
		(1) Hypochlorhydria and Hyperchlorhydria	
		(2) Hypokalemia and Hyperkalemia	
		(3) First order reaction and Second order reaction	
8	Answer the following:		
	(1)	How the radioactivity measured? Write a note on Gieger Muller counter.	7
	(2)	Write the assay principle of following compounds.	6
		(1) Boric acid	
		(2) Copper sulphate	
		(3) Hydrogen peroxide	
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